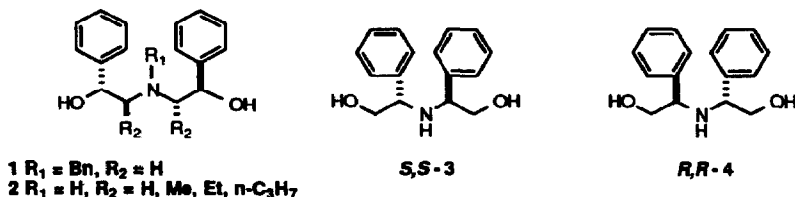


A Stereoselective Concise Synthesis of C_2 - and *meso*-Aminodiols from (*R*)-Phenylglycinol

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Summary : Oxazolidines derived from (*R*)-phenylglycinol can undergo diastereoselective vinyl- and aryl addition to give, after oxidative cleavage, *meso*- and C_2 -diethanol amines.

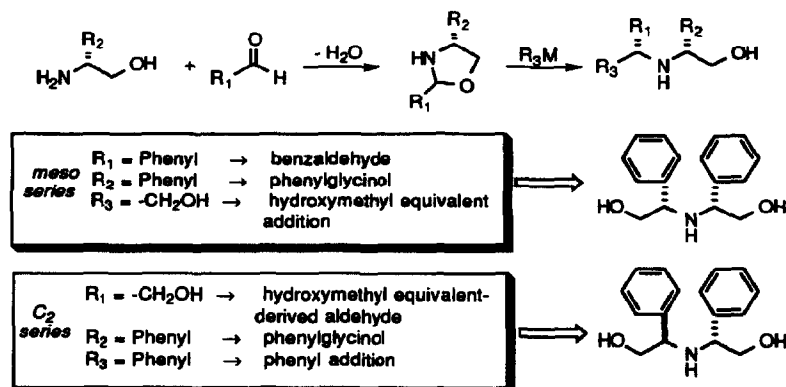
Recently, aminodiols have been employed as tridentate chiral modifiers in the asymmetric reduction of prochiral ketones.¹ These tridentate ligands offer a third binding site in the central nitrogen and provide an opportunity for C_2 -symmetry. Aminodiol **1**, prepared from (*R*)-phenyloxirane and benzylamine, has been shown to be a C_2 -scaffold for a bis-phosphinoester that serves as ligand for a catalytic Pd(0)-mediated asymmetric allylic alkylation.² C_2 -aminodiols **2**, obtained from an optically active *O*-protected cyanohydrin,³ have been utilized as key building blocks in the construction of chiral diaza-18-crown-6 derivatives.⁴ A C_2 -scaffold has also been implicated as the key structural feature responsible for the binding of inhibitors of HIV-1 protease.⁵



A recent report from Durst and co-workers utilized the reaction of (*R*)-pantolactone esters of α -halocarboxylic acids with optically active α -amino esters in the preparation of a new class of C_2 -diethanol amines, **3**.⁶ This report prompts us to describe our route to C_2 -diethanol amines **4** from (*R*)-phenylglycinol.

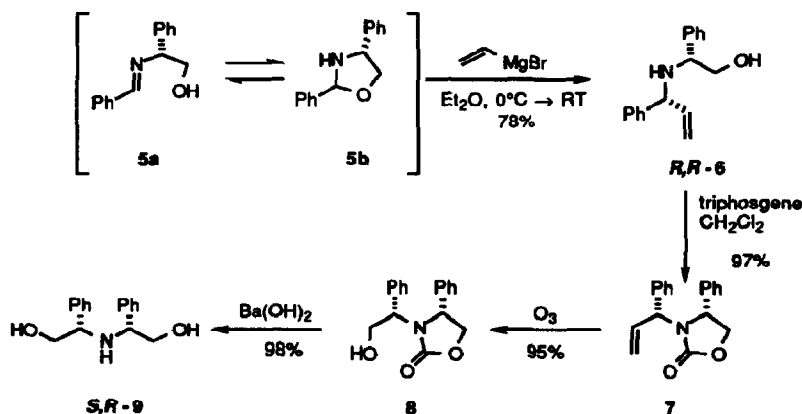
As demonstrated recently⁷, oxazolidines derived from optically active amino alcohols have been shown to undergo chelation-controlled diastereoselective alkylations. We envisioned that symmetric diethanol amines derived from amino alcohols could be prepared via this methodology (Scheme 1). Both the *meso*- and C_2 -diethanol amines can be accessed by appropriate choice of aldehyde and organometallic reagent. Addition of a hydroxymethyl equivalent (vinylmagnesium bromide)⁸ to the oxazolidine derived from benzaldehyde and phenyl glycinol should lead, after ozonolytic cleavage and reduction, to the *meso*-diethanol amine. Likewise, diastereoselective phenyl addition to the oxazolidine derived from a hydroxymethyl equivalent-containing aldehyde

(cinnamaldehyde), and phenyl glycinol would produce, after ozonolytic cleavage and reduction, the *C*₂-diethanol amine.



Scheme 1

To demonstrate this protocol, oxazolidinone **5⁹**, prepared from benzaldehyde and (*R*)-phenyl glycinol underwent chelation controlled vinyl addition upon treatment with vinylmagnesium bromide (3 equiv.) to give **6** in a diastereoselectivity of 11:1 by ¹H NMR (Scheme 2).¹⁰

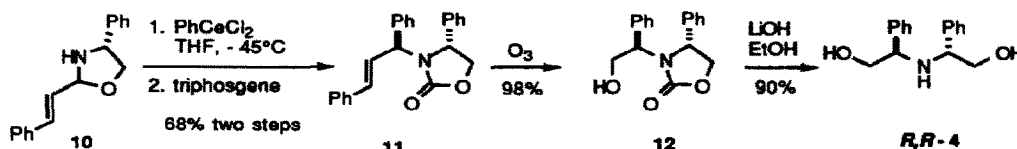


Scheme 2

Transformation of the amino alcohol to the oxazolidinone **7** using triphosgene (0.4 equiv.)¹¹ followed by ozonolysis afforded hydroxy-oxazolidinone **8** after reductive workup with NaBH₄. Hydrolysis of the oxazolidinone with barium hydroxide¹² gave the *meso*-diethanol amine **9** in

good yield.¹³ The stereochemistry of **8** was confirmed by single crystal x-ray determination which supports chelation controlled addition to **5**.

The C_2 -symmetric aminodiols **4** was prepared as follows: oxazolidinone **10**, prepared from *trans*-cinnamaldehyde and (*R*)-phenyl glycinol¹⁴ underwent complete diastereoselective phenyl addition on treatment with phenylcerium dichloride (Scheme 3).¹⁵



Scheme 3

Formation of the oxazolidinone **11** (triphosgene, CH_2Cl_2) was once again followed by ozonolysis and reductive workup (NaBH_4) to give hydroxy-oxazolidinone **12** in near quantitative yield.¹⁶ Finally, hydrolysis of the latter with ethanolic lithium hydroxide then furnished C_2 -diethanol amine **4** in good yield.¹⁷

In conclusion we have described an expedient, high yielding routes to the (*R*)-phenyl glycinol-derived *meso*- and C_2 -diethanol amines. The present route to aminodiols **9** and **4** can be run on multigram scale. The synthetic application of these type of tridentate ligands is currently under investigation.

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References

- 1 Aminodiols as chiral modifiers for: a) LAH reduction, de Vries, E. F.J.; Brussee, J.; Kruse, C. G.; van der Gen, A. *Tetrahedron Asymmetry* **1994**, *5*, 377. b) Catalytic asymmetric Meerwin-Pondorf reduction, Evans, D. A.; Nelson, S. G.; Gagne, M. R.; Muci, A. R. *J. Am. Chem. Soc.* **1993**, *115*, 9800.
- 2 Trost, B. M.; Van Vranken, D. L.; Bingel, C. *J. Am. Chem. Soc.* **1992**, *114*, 9327.
- 3 Brussee, J.; van der Gen, A. *Recl. Trav. Chim. Pays-Bas.* **1991**, *110*, 25.
- 4 de Vries, E. F.J.; Steenwinkel, P.; Brussee, J.; Kruse, C. G.; Van der Gen, A. *J. Org. Chem.* **1993**, *58*, 4315.
- 5 Madhusoodan, V. H.; Bhat, T. N.; Kempf, D. J.; Baldwin, E. T.; Liu, B.; Gulnik, S.; Wideburg, N. E.; Norbeck, D. W.; Appelt, K.; Erickson, J. W. *J. Am. Chem. Soc.* **1994**, *116*, 847, and references cited therein.

- 6 Koh, K.; Ben, R. N.; Durst, T. *Tetrahedron Lett.* **1994**, *35*, 375.
- 7 a) Higashiyama, K.; Inoue, H.; Takahashi, H. *Tetrahedron* **1994**, *50*, 1083 and earlier references cited. b) Pridgen, L. N.; Wu, M.-J. *Synlett* **1990**, 636. c) Pridgen, L. N.; Wu, M.-J. *J. Org. Chem.* **1991**, *56*, 1340. d) Miao, C. K.; Sorcek, R.; Jones, P.-J. *Tetrahedron Lett.* **1993**, *34*, 2259. e) Bocoum, A.; Savoia, D.; Umani-Ronchi, A. *J. Chem. Soc., Chem. Commun.* **1993**, 1542.
- 8 To our knowledge addition of sp² nucleophilic organometallic species to chiral hydroxy-aldimines (oxazolidines) has not been reported.
- 9 It was found that these oxazolidines of phenyl glycinol exist predominately in solution as their hydroxy-imine tautomers as demonstrated by the appropriate resonances in their ¹H and ¹³C spectra.
- 10 All new compounds displayed spectroscopic data consistent with their structural assignments.
- 11 Correa, A.; Denis, J.-N.; Greene, A. E. *Synth. Commun.* **1991**, *21*, 1.
- 12 Hoskins, W. M.; Crout, D. H. G. *J. Chem. Soc., Perkin Trans.* **1977**, *1*, 538.
- 13 Purified by flash chromatography on Amicon silica gel using 50% EtOAc/hexanes → EtOAc and isolated as a colorless crystalline solid (mp = 106 -107 °C). ¹H-NMR (CDCl₃) δ : 7.22 (m, 5H), 3.84 (dd, J = 4.72 and 6.63 Hz, 1H), 3.78 (dd, J = 4.72 and 10.90 Hz, 1H), 3.64 (dd, J = 6.63 and 10.90 Hz), 2.66 (br s, 1.5H). ¹³C-NMR (CDCl₃) δ : 140.9 (s), 128.6 (d), 127.6 (d), 127.2 (d), 66.0 (t), 61.8 (d). Anal. Calcd. for C₁₆H₁₉NO₂: C, 74.68; H, 7.44; N, 5.44. Found : C, 74.59; H, 7.40; N, 5.39. [α]_D²⁵ = 0°.
- 14 Pridgen, L. N.; Mokhallalati, M. K.; Wu, M.-J. *J. Org. Chem.* **1992**, *57*, 1237.
- 15 Imamoto, T.; Takiyama, N.; Nakamura, K.; Hatajima, T.; Kamiya, Y. *J. Am. Chem. Soc.* **1989**, *111*, 4392.
- 16 Purified by flash chromatography on Amicon silica gel using 25 → 50% EtOAc/hexanes.
- 17 Crude (*R,R*)-4, purified by flash chromatography on Amicon silica gel using 50% EtOAc/hexanes → EtOAc was isolated as a foam. HCl salt white powder (Mp > 250 °C d). ¹H-NMR (CDCl₃) δ : 7.27 (m, 5H), 3.60 (m, 3H), 2.55 (br s, 1.5 H). ¹³C-NMR (CDCl₃) δ : 139.9 (s), 128.7 (d), 127.7 (d), 127.4 (d), 67.2 (t), 61.0 (d). Anal. Calcd. for C₁₆H₂₀ClNO₂ : C, 65.41; H, 6.86; N, 4.76. Found : C, 65.22; H, 6.91; N, 4.68. [α]_D²⁵ = 193.4 ° (c 1.4, CH₂Cl₂). Lit.⁶ [α]_D²⁵ = 133° (*S,S* enantiomer). Chiral HPLC analysis showed (*R,R*) as a single peak on Chiralcel OD column (rt = 42.16 min., 96:4 IPA/hexanes @ flow rate of 1 mL/min). Verification of ee was accomplished by comparison with baseline-separated racemic-4.

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